

Product Data Sheet

AMBERLITE[™] HPR4200 CI Ion Exchange Resin

Uniform Particle Size, Gel, Strong Base Anion Exchange Resin for Industrial Demineralization Applications

Description	AMBERLITE [™] HPR4200 CI Ion Exchange Resin is a high-quality resin for use in industrial demineralization applications when high performance and cost-effective operation is required. The chemical properties and particle size of the resin have been balanced to combine excellent operating capacity with low pressure drop, while reducing chemical regenerant and rinse water usage.			
	AMBERLITE HPR4200 CI is compatible with all system technologies; it has the flexibility to be used in lead single or layered anion bed and in mixed bed polishers, allowing users to inventory only one strong base anion resin for their demineralization needs. In mixed bed applications, the light color of this anion resin is designed to allow easy visual distinction from the dark-colored cation resin following backwash separation.			
	AMBERLITE HPR4200 CI is designed for use in single bed or layered bed systems when paired with AMBERLITE [™] HPR9600 or AMBERLITE [™] HPR9500 Ion Exchange Resins. In mixed bed applications AMBERLITE [™] HPR4200 OH Ion Exchange Resin is recommended, but AMBERLITE HPR4200 CI may be used if chloride-form is preferred by the user.			
Resin Pairings	 Recommended pairing in industrial demineralization applications: AMBERLITE™ HPR1200 H Ion Exchange Resin (gel) – for mixed bed AMBERLITE™ HPR1300 H Ion Exchange Resin (gel) – for mixed bed AMBERLITE™ HPR9500 Ion Exchange Resin (macroporous) – for layered bed AMBERLITE™ HPR9600 Ion Exchange Resin (macroporous) – for layered bed 			
Applications	 Demineralization Ideally when treating water with: High percentage of silica When the treatment goal is: Removal of strong and weak acids Lowest silica leakage Mixed bed polishing 			

System Designs	Compatible with all system techn • Co-current • Counter-current / Hold-dow • Layered beds • Packed beds • Mixed beds	
Historical Reference	AMBERLITE™ HPR4200 CI Ion DOWEX MARATHON™ 4200 CI	Exchange Resin has previously been sold as I lon Exchange Resin.
Typical Physical	Physical Properties	
and Chemical	Copolymer	Styrene-divinylbenzene
Properties**	Matrix	Gel
•	Туре	Strong base anion, Type I
	Functional Group	Trimethylammonium
	Physical Form	Yellow, translucent, spherical beads
	Chemical Properties	· · · · · · · · · · · · · · · · · · ·
	Ionic Form as Shipped	CI-
	Total Exchange Capacity	≥ 1.30 eq/L (Cl [−] form)
	Water Retention Capacity	49.0 – 55.0% (Cl ⁻ form)
	Particle Size	
	Particle Diameter §	650 ± 50 μm
	Uniformity Coefficient	≤ 1.25
	< 300 µm	≤ 0.3%
	> 850 μm	≤ 5.0%
	Stability	
	Whole Uncracked Beads	≥ 90%
	Swelling	$CI^- \rightarrow OH^-$: 20%
	Density	
	Particle Density	1.07 g/mL

§ For additional particle size information, please refer to the <u>Particle Size Distribution Cross Reference Chart</u> (Form No. 177-01775).

670 g/L

Shipping Weight

Suggested Operating Conditions**

Temperature Range	
OH⁻ form ‡	5 – 60°C (41 – 140°F)
Cl [−] form	5 – 100°C (41 – 212°F)
pH Range	
Service Cycle	1 – 14
Stable	0 – 14

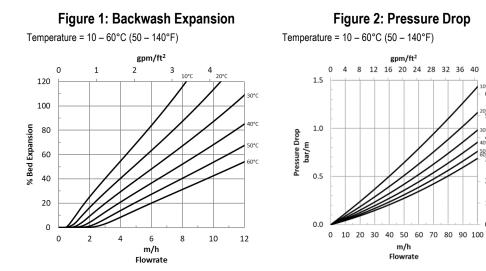
[‡] Operating at elevated temperatures, for example above 60 - 70°C (140 - 158°F), may impact resin life. Contact our technical representative for details.

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for mixed beds (Form No. 177-03705) or separate beds (Form No. 177-03729) in water treatment, please refer to our Tech Facts.

Hydraulic Characteristics

Estimated bed expansion of AMBERLITE™ HPR4200 CI Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AMBERLITE HPR4200 CI as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water.



°°C 6.0

)°C 5.0

4.0 psi/ft

3.0³³⁸⁸

2.0

1.0 0.0

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WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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